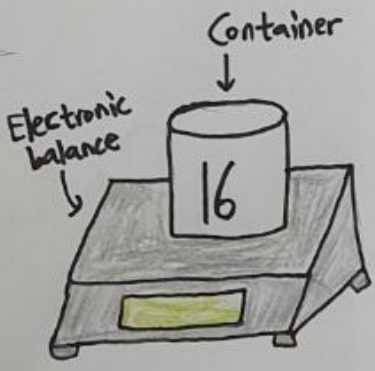
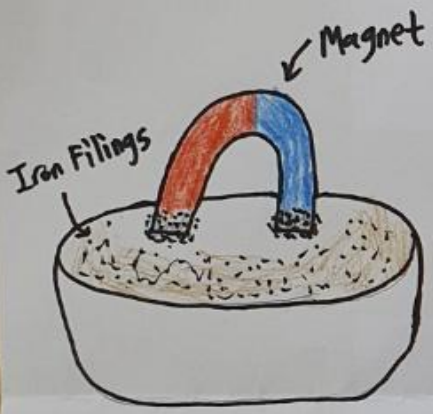


# Separation of Mixtures Lab Flowchart



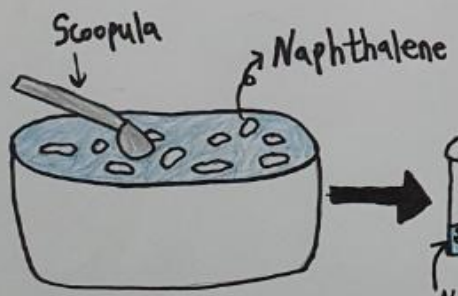
1. Record the **number** assigned to the container (Recording)
2. Measure the **total mass** of the mixture initially including the container using the **electronic balance** (Measuring)
3. Empty the container and measure the mass of the empty container to obtain the exact mass of the mixture (Measuring)



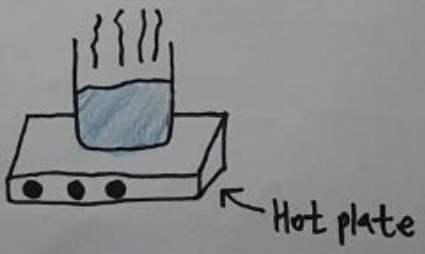
4. Use a **magnet** to separate the **Iron Filings** from the mixture (Magnetism)
5. Measure the **mass** of the **Iron Filings** (Measuring)



6. Pour **water** into the mixture (Solubilizing)
7. Let the salt totally **dissolve** in the water by stirring the mixture using a **stir bar** (Solubilizing)
8. Scoop up the substance floating on top of the solution, which is **naphthalene**, using a **scoopula** (Sorting)
9. Use the **gravimetric filtration system** to separate the saltwater and sand with a funnel and a filter paper and naphthalene and saltwater (gravimetric filtration)



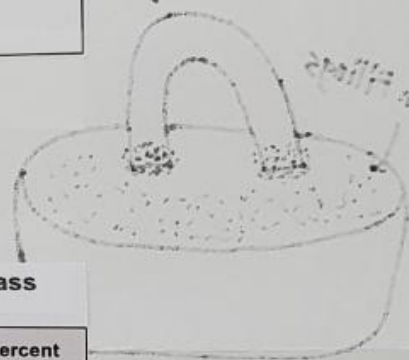
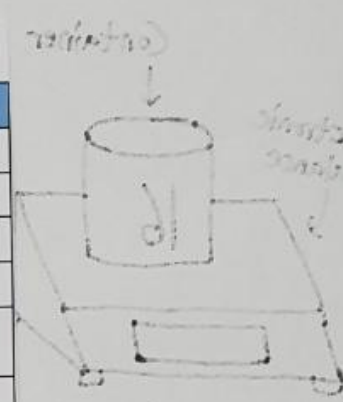
10. Measure the **mass of sand** and the **mass of naphthalene** with an electronic balance (Measuring)
11. Heat the saltwater until the water completely **evaporates** using a hot plate (Evaporation)
12. Measure the **mass of salt** remaining using the electronic balance (Measuring)
13. Check if the sum of each substance is equal to the mass of the whole mixture measured initially (Confirmation, check)



Data Table attached on the back side

### Expected Mass of the Substances from the Separation Lab

Substances	Measured Mass (g)
Iron Filings	3.01
Naphthalene	0.49
Sand	4.59
Sodium Chloride (NaCl)	2.21
Initial mass of the mixture including the container	17.08
Mass of the empty container	6.78
Initial mass of the mixture (before experiment)	10.3
Final mass of the mixture (after experiment)	10.3



### Average Percent Error Between the Expected Mass and Actual Mass

Substances	Measured Mass (g)	Actual Mass (g)	Percent Error (%)	Average Percent Error (%)
Iron Filings	3.010	2.930	2.730	13.560
Naphthalene	0.490	0.734	33.243	
Sand	4.590	4.693	2.195	
Sodium Chloride	2.210	1.904	16.071	



(initial) including the container  
the empty container  
(initial) mixture (Mixture)

the mixture using  
and while is  
step in electronic balance  
to separate the saltwater and sand  
(mixture) (mixture)  
with an electronic

completely  
using the electronic balance (mixture)  
of the mixture (mixture) initially (mixture) (mixture) check

(Mixture) of the (Mixture)

(mixture) (mixture)

1. Measure the (mixture) (mixture)  
2. Measure the (mixture) (mixture)  
3. Check if the sum of each substance is equal to the mass

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# Data Tables

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